

SYR - FIGURE 4-2

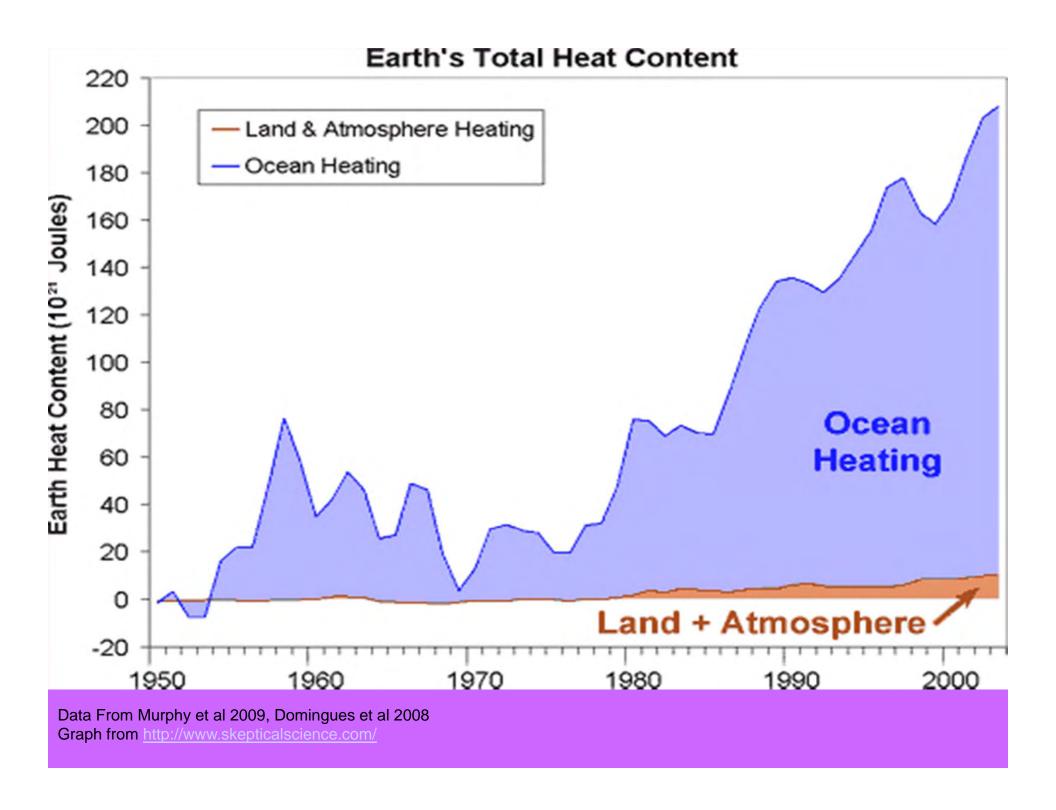




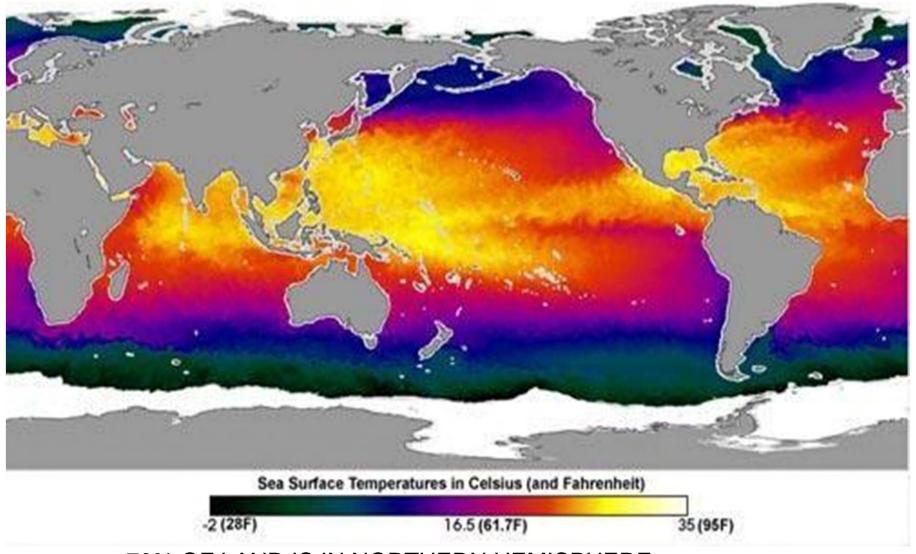
Comparison of the heat balance of the climate system

Levitus et al (2001). Science Vol. 292, pp. 268.

Component of the climate system and source of data	Time period of change	Observed or estimated change	Heat content increase or total heat of fusion	%
World ocean	1955-1996	Observed temperature increase	18.2 X 10 ²² J	90%
Global atmosphere	1955-1996	Observed temperature increase	6.6 X 10 ²¹ J	3
Decrease in the mass of continental glaciers	1955-1996	-	8.1 X 10 ²¹ J	4
Decrease in Antarctic sea ice extent	1950s-1970s	Estimated 311-km reduction in sea ice edge	3.2 X 10 ²¹ J	1
Mountain glacier decrease	1961-1997	3.7 X 10 ³ km decrease in mountain glacier ice volume	1.1 X 10 ²¹ J	.5
Decrease in Northern Hemisphere sea ice extent	1978-1996	Areal change based on satellite measurements	4.6 X 10 ¹⁹ J	.02
Decrease in Arctic perennial sea ice volume	1950s- 1990s	40% decrease in sea ice thickness	2.4 X 10 ¹⁹ J	.01

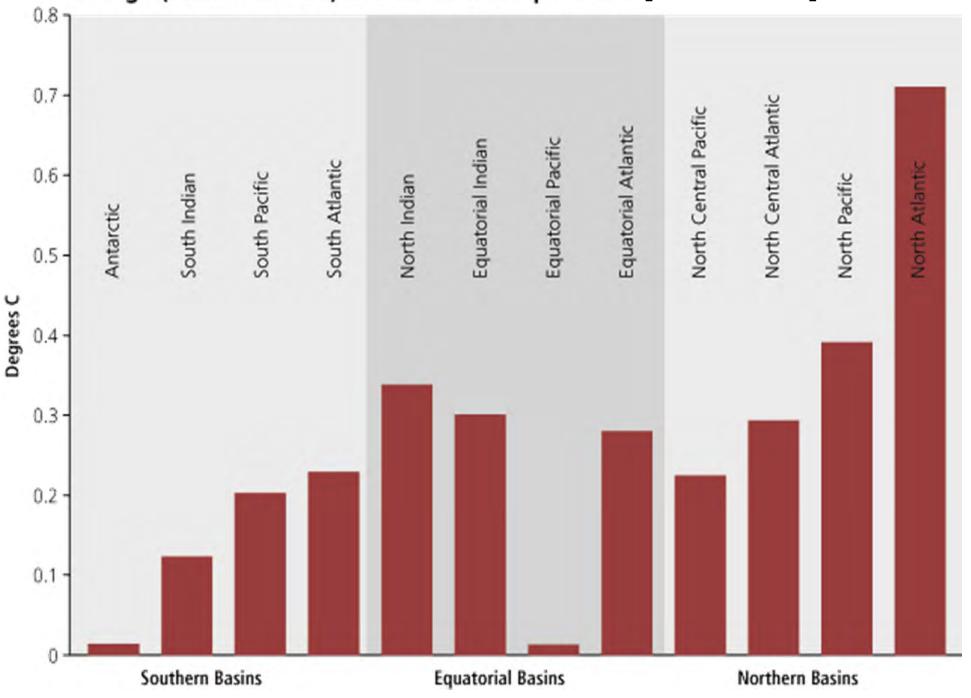


OCEAN SURFACE TEMPERATURES



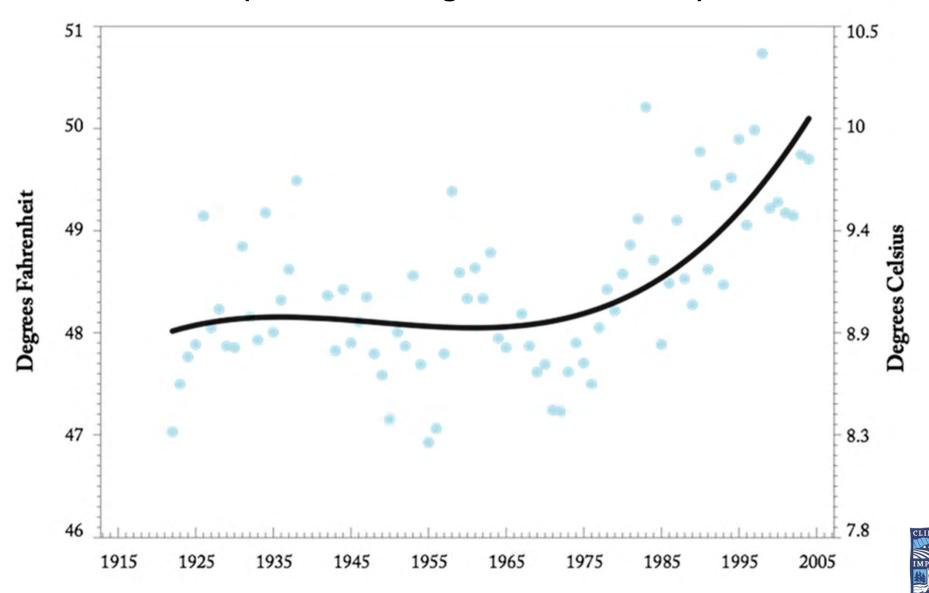
70% OF LAND IS IN NORTHERN HEMISPHERE

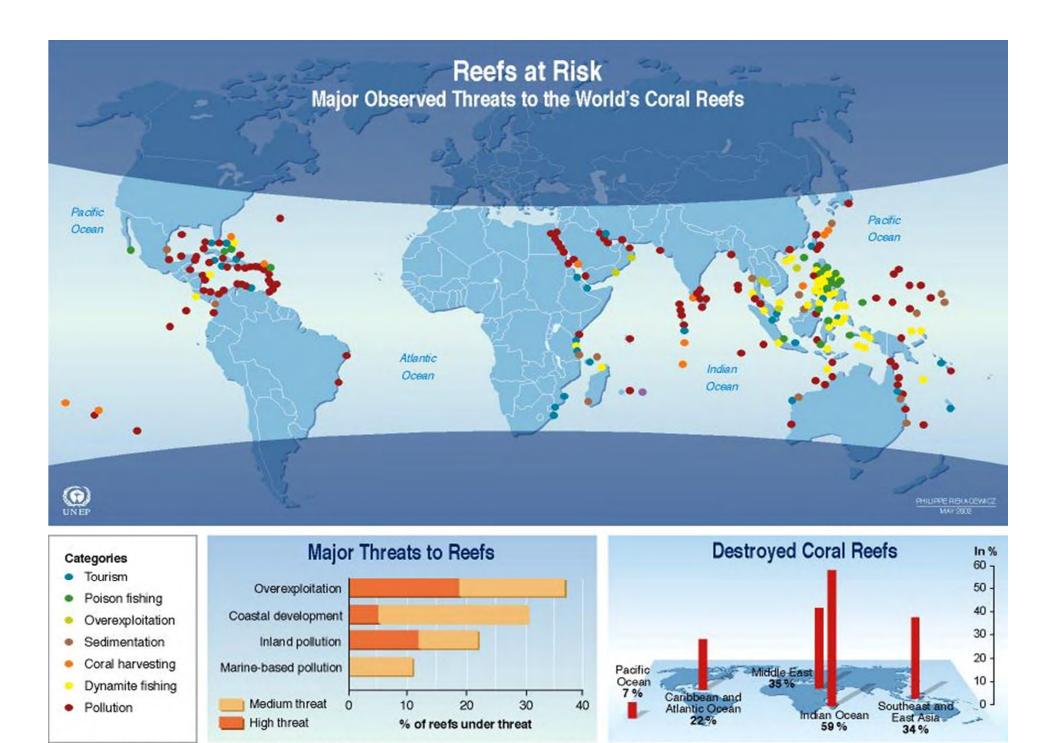
Change (SeaWiFS-CZCS) Sea Surface Temperature [1979 - 2002]



Sea Surface Temperature

(Race Rocks lighthouse, Victoria)





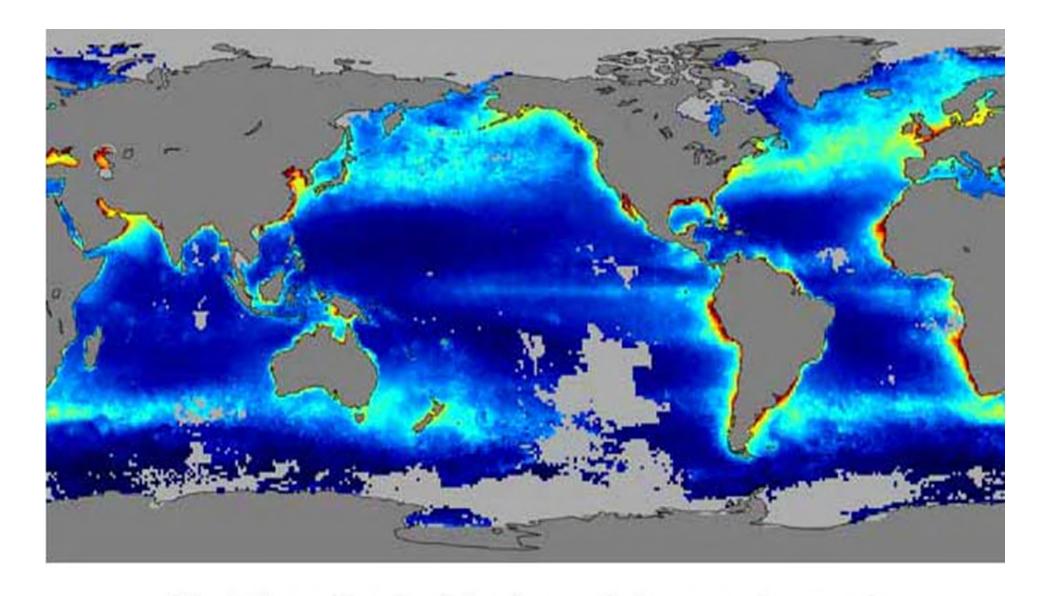
Source: Bryant et al., Reefs at Risk; a Map-Based Indicator of Threats to the World's Coral Reefs, World Resources Institute (WRI), Washington DC, 1998.



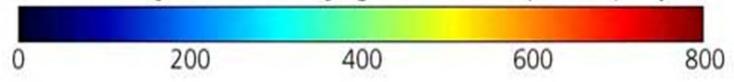
BLEACHING OF CORAL REEFS BY OCEAN TEMPS > 85deg



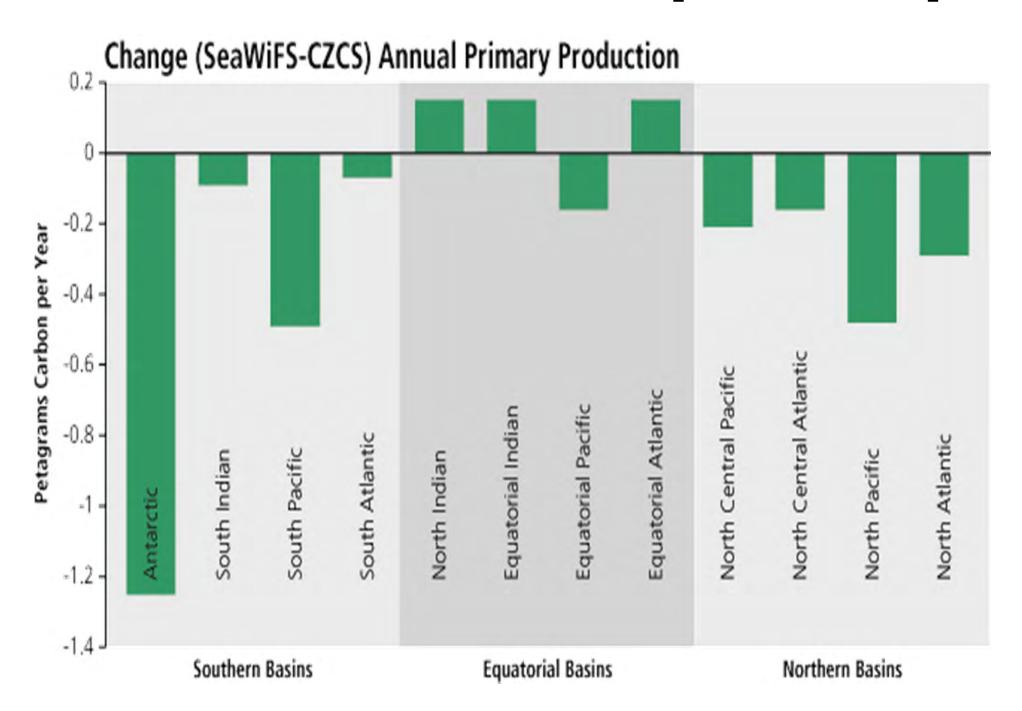




Net Primary Productivity (grams Carbon per m² per year)



CHANGE IN OCEAN NPP [1979 - 2002]



Ocean acidification

 CO₂ is corrosive to the shells and skeletons of many marine organisms

Corals



Calcareous plankton

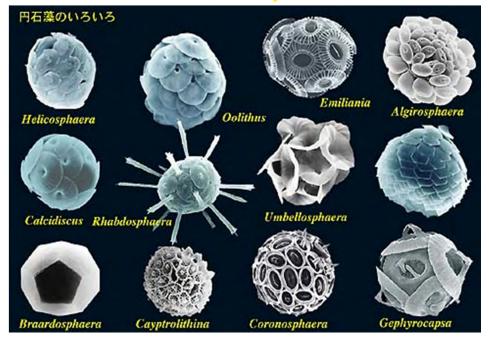


Photo: Missouri Botanical Gardens

http://www.biol.tsukuba.ac.jp/~inouye

Ocean Acidification

Over the last 200 years, about 50% of all CO₂ produced on earth has been absorbed by the ocean. (Royal Society 6/05)

Dissolves in sea water

 $CO_2 + H_2O \longrightarrow HCO_3^- + H^+$

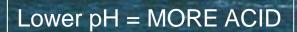
(ACID)

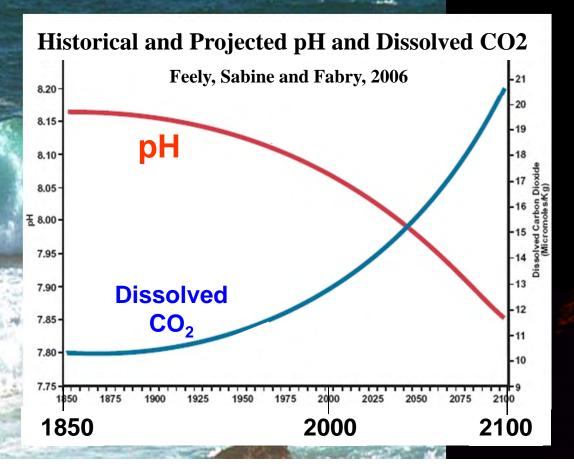
Water becomes more acidic.

Remains in the atmosphere (greenhouse gas)

Warming: The Greatest Threat © 2006 Deborah L. William

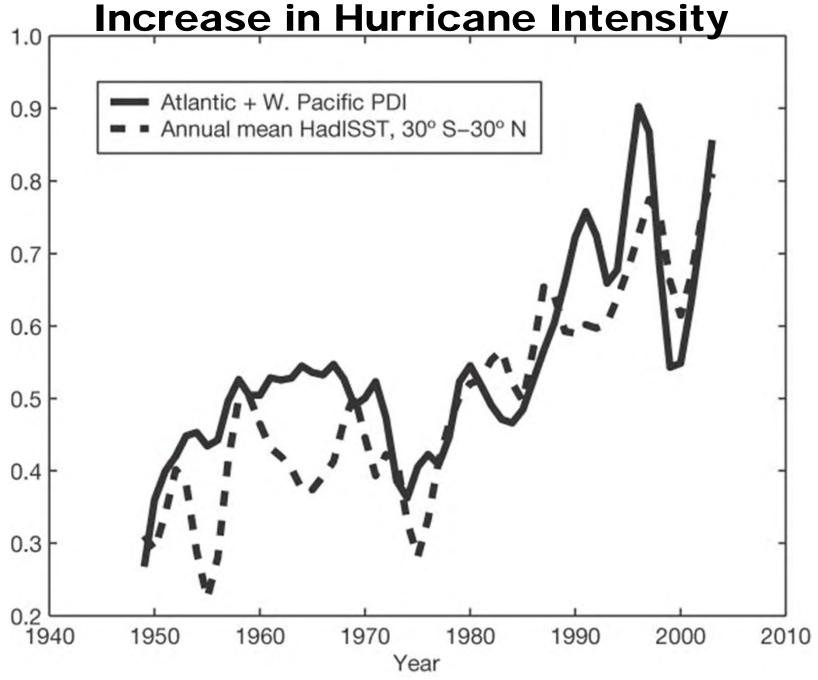
Ocean Acidification





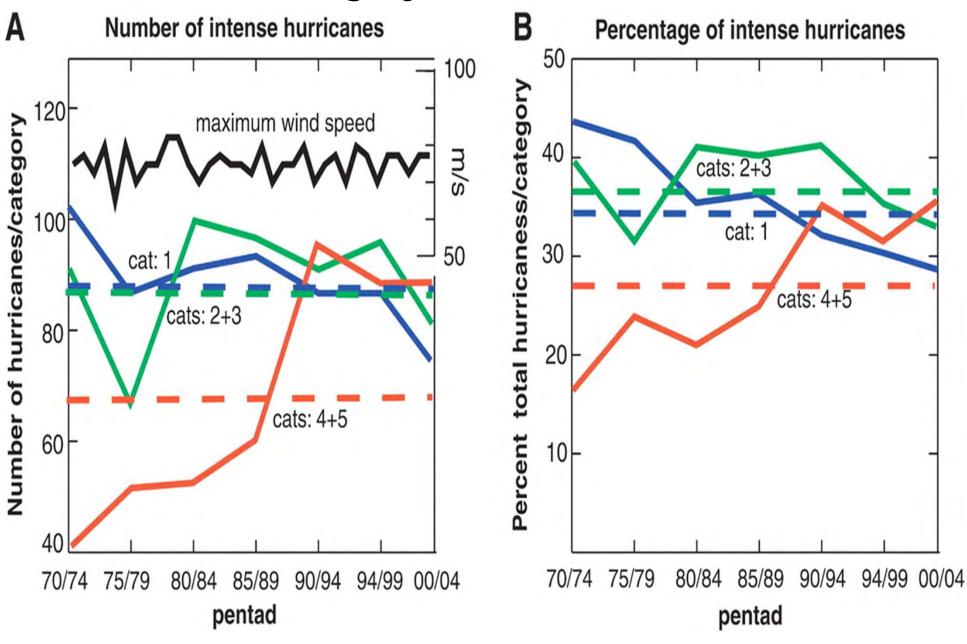
- Since 1850, ocean pH has decreased by about 0.1 unit (30% increase in acidity). (Royal Society 2006)
- At present rate of CO₂ emission, pH predicted to increase by 0.4 units (3-fold increase in H ions) by 2100.
- Carbonate ion concentrations decrease.





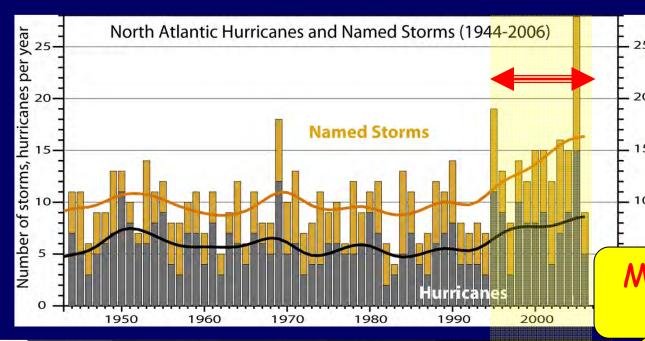
PDI = Potential Destructiveness Index

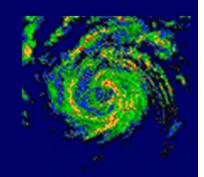
Increase in Category 4-5 Hurricanes 1970 - 2004



Webster et al, Science, Sept 16 2005

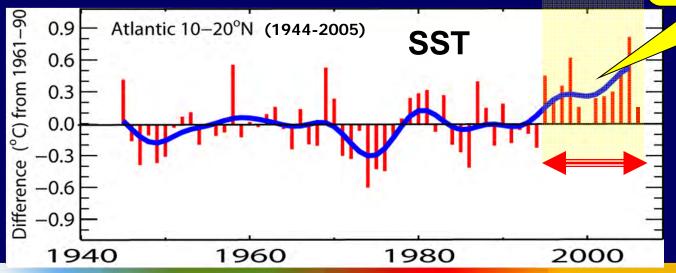
North Atlantic hurricanes have increased with SSTs



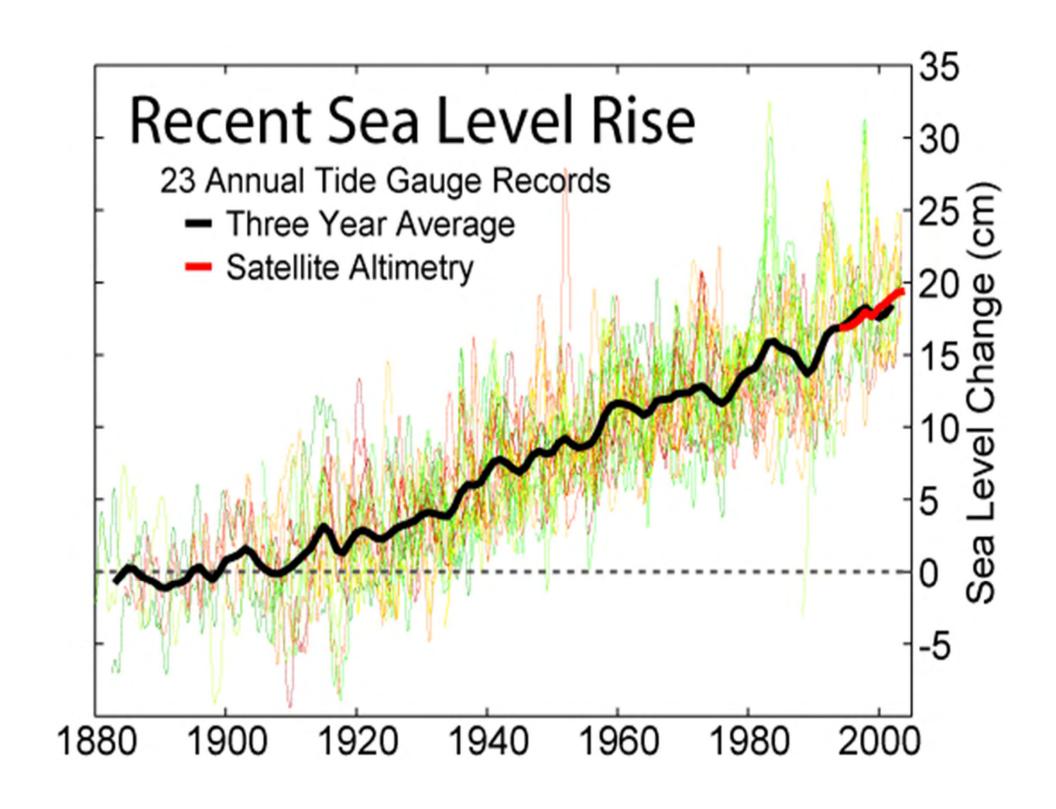


N. Atlantic hurricane record best

Marked increase after 1994



Global number and percentage of intense hurricanes is increasing



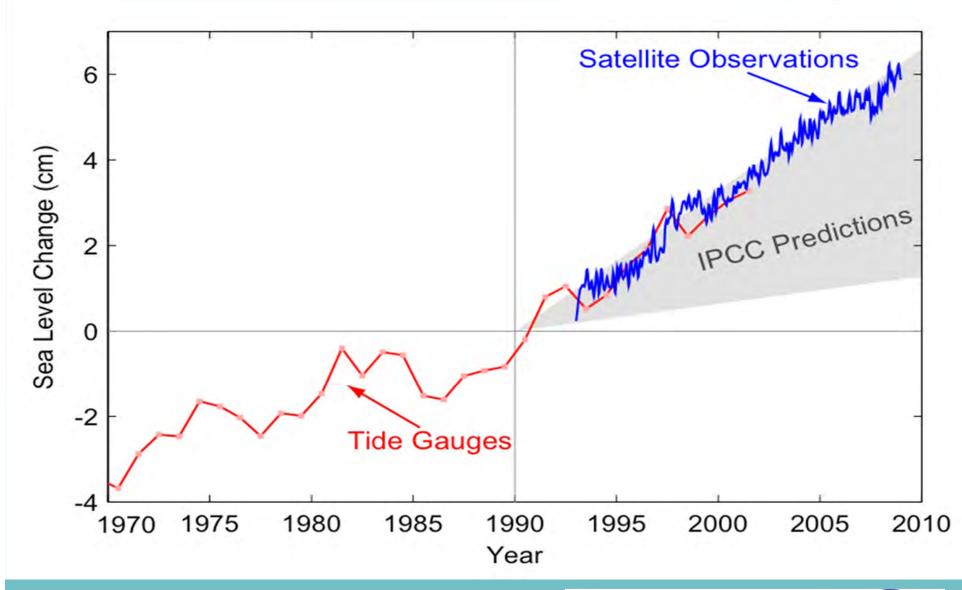


Figure 16: Sea-level change 1970-2010



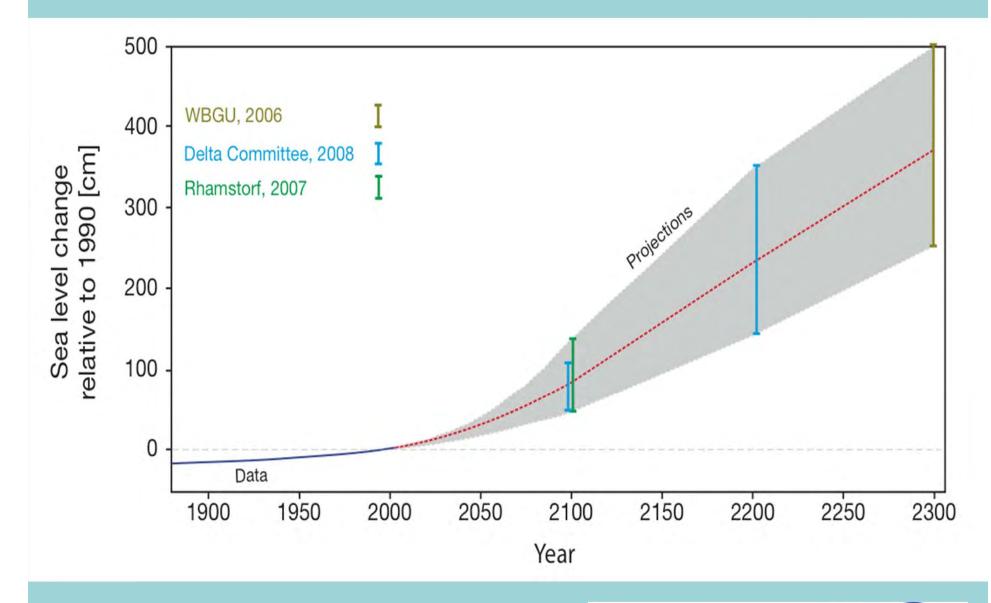
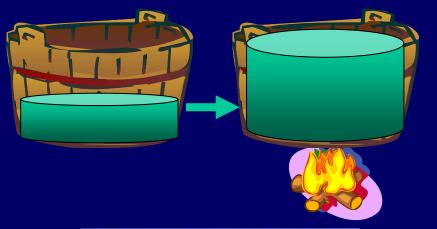


Figure 17: Past and future sea-level projections



Sea-level Rise Projections Include:



 ocean expansion resulting from increased water temperatures;



 meltwater runoff from mountain glaciers around the world; and



•a contribution due to increased ice flow from Greenland and Antarctica at the rates observed for 1993-2003.

Source: IPCC Climate Change 2007: The Physical

Science Basis—Summary for Policymakers.

Sea-level Rise Projections DO NOT Include:





- Ice sheet instability
- Carbon dioxide uptake changes

IPCC: "Larger values cannot be excluded, but understanding of these effects is too limited to assess their likelihood or provide a best estimate or an upper bound for sea-level rise."

Pensacola Panama City Pensacola Panama City Pensacola Panama City Pensacola Gulf of Mexico Cape Canaveral Saint Petersburg FLORIDA Fort Lauderdale Miami Weiss and Overpeck The University of Arizona Panama City Tallahassee Gainsville Gainsville Gainsville Petersburg FLORIDA Fort Lauderdale Miami Key West

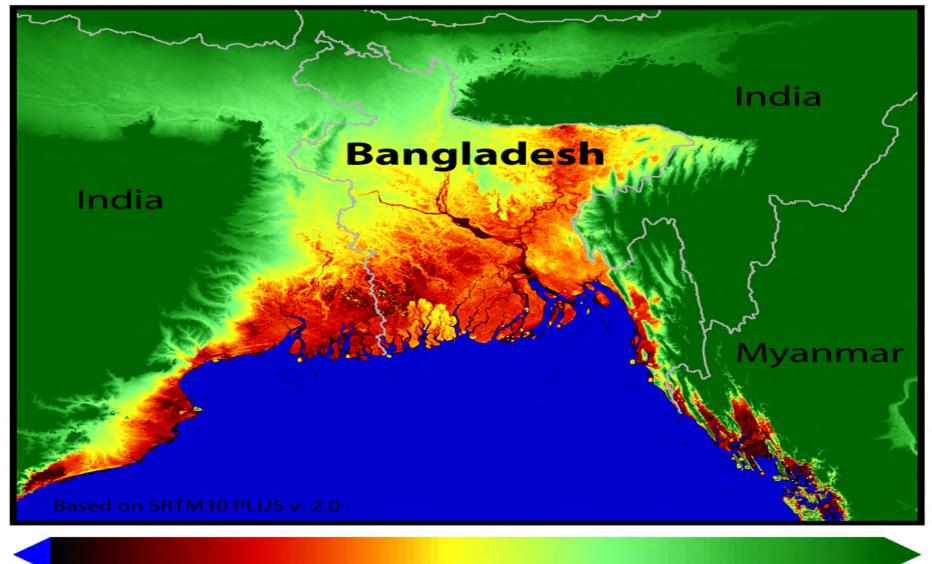


Threshold risks:

Some models do suggest that sustained warming between 2-7°F above today's global average temperature would initiate irreversible melting of the Greenland ice sheet—which could ultimately contribute about 23 *feet* to sea-level rise.

Source: IPCC *Climate Change 2007: The Physical Science Basis*—Summary for Policymakers.

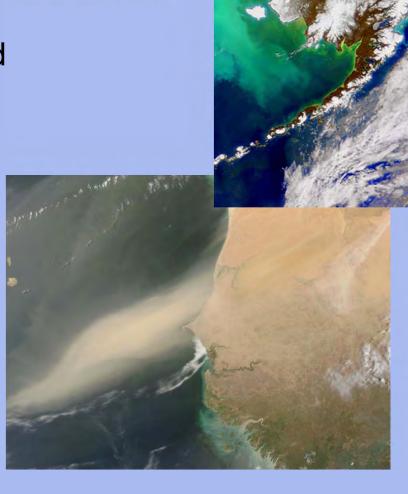
Sea Level Risks - Bangladesh





Iron in the Oceans

- Sources of naturally occurring iron
 - Volcanic coastal shelves
 - Dust in blown in from land
 - Upwellings
- Role in ecosystems
 - Key nutrient that helps plants take up nitrogen





"Give me a few oil tankers full of iron, and I'll give you an ice age."

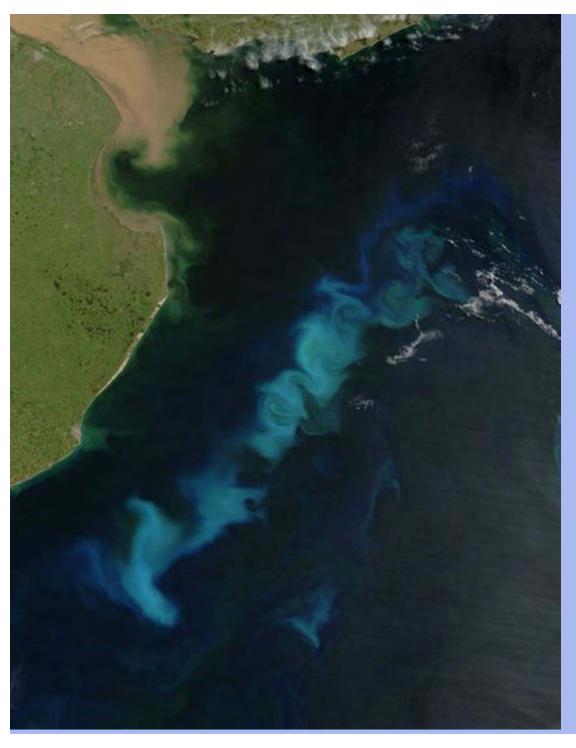
- John Martin, WHOI Scientist

Cashing in on Carbon Offsets

FERTILIZING OCEANS FOR FUN AND PROFIT A boat goes out to the middle of the ocean and sprinkles micron-size iron shavings onto the surface of the water. The iron shavings stimulate the growth of photosynthesizing algae, which uses sunlight to draw carbon dioxide out of the air. 3 When the algae dies, it bleaches and sinks into the sea. If it sinks to 1,000 feet, the carbon should be trapped for decades. PHOTO-ILLUSTRATION: MAGICTORCH

- Climos taking over with \$3.5 million in funding
- Planktos bottom up after Galapagos proposal





Will the Carbon stay sequestered long enough to help?

NOPE!

- CLIMATE CHANGE CURE?: By running the flue gas from Moss Landing's mammoth smokestacks through ocean water, a new company can make cement from carbon dioxide pollution.
- The turbines at Moss Landing power plant on the California coast burn through natural gas to pump out more than 1,000 megawatts of electric power. The 700-degree Fahrenheit (370-degree Celsius) fumes left over contain at least 30,000 parts per million of carbon dioxide (CO2)—.

Today, this flue gas wafts up and out of the power plant's enormous smokestacks, but by simply bubbling it through the nearby seawater, a new California-based company called Calera says it can use more than 90 percent of that CO2 to make something useful: cement.

It's a twist that could make a polluting substance into a way to reduce greenhouse gases. Cement, which is mostly commonly composed of calcium silicates, requires heating limestone and other ingredients to 2,640 degrees F (1,450 degrees C) by burning fossil fuels and is the third largest source of greenhouse gas pollution in the U.S., according to the U.S. Environmental Protection Agency. Making one ton of cement results in the emission of roughly one ton of CO2—and in some cases much more.

Calera's process takes the idea a step forward by storing the CO2 in a useful product. The U.S. used more than 122 million metric tons of Portland cement in 2006, according to the Portland Cement Association (PCA), an industry group, and China used at least 800 million metric tons.

The Calera process essentially mimics <u>marine cement</u>, which is produced by coral when making their shells and reefs, taking the calcium and magnesium in seawater and using it to form carbonates at normal temperatures and pressures. "We are turning CO2 into carbonic acid and then making carbonate," Constantz says. "All we need is water and pollution."